

5

$$V_0 = 2,35 \text{ dm}^3 = 2,35 \cdot 10^{-3} \text{ m}^3$$

$$T = 31^\circ \text{C} = 304,15 \text{ K}$$

$$M_{\text{mol}} = 28 \frac{\text{g}}{\text{mol}} = 28 \cdot 10^{-3} \frac{\text{kg}}{\text{mol}}$$

$$p = 965 \text{ mbar} = 0,965 \text{ bar} = 0,965 \cdot 10^5 \text{ Pa}$$

$$(a) \quad pV = nRT$$
$$n = \frac{pV}{RT} = \frac{0,965 \cdot 10^5 \text{ Pa} \cdot 2,35 \cdot 10^{-3} \text{ m}^3}{8,31446 \frac{\text{J}}{\text{molK}} \cdot 304,15 \text{ K}}$$

$$n = 0,0896754 \text{ mol}$$

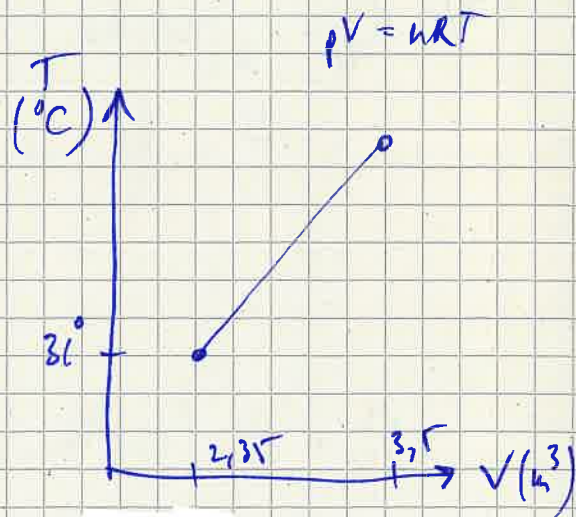
$$m = n \cdot M_{\text{mol}} = 0,0896754 \text{ mol} \cdot 28 \frac{\text{g}}{\text{mol}}$$

$$m = 2,510911 \text{ g}$$

$$m = 2,5 \text{ g}$$

(b)

$$pV = nRT$$
$$p, R, T = \text{const}$$
$$\rightarrow p \sim T$$



$$(c) \quad W = - \int_{V_0}^{V_f} p(V) dV = - \int_{V_0=2,35 \text{ dm}^3}^{V_f=3,5 \text{ dm}^3} 965 \text{ mbar} dV$$

$$W = -965 \text{ mbar} (3,5 \text{ dm}^3 - 2,35 \text{ dm}^3)$$

$$W = -1,10975 \text{ bar dm}^3$$

$$W = -1,1 \cdot 10^2 \text{ J}$$

$$\text{bar dm}^3 = 10 \frac{\text{kg}}{\text{m}^2} \cdot 10^{-3} \text{ m}^3$$
$$= 10^2 \text{ J}$$

⑥

$S_1: p_1 = 2,2 \text{ atm} = 2,22926 \cdot 10^5 \text{ Pa}$
 $T_1 = 150 \text{ K}$
 $V_1 = 6,5 \text{ m}^3$

$V = \text{const}$ ↓

$S_2: V_2 = 6,5 \text{ m}^3$
 $T_2 = 240 \text{ K}$
 $p_2 =$

$T = \text{const}$ ↓

$S_3: V_3 = 12 \text{ m}^3$
 $T_3 = 240 \text{ K}$
 $p_3 =$

(a) $S_1 \rightarrow S_2$ (isochor): $\frac{p}{T} = \text{const}$

$$\frac{p_1}{T_1} = \frac{p_2}{T_2} \rightarrow p_2 = p_1 \cdot \frac{T_2}{T_1} = 2,2 \text{ atm} \cdot \frac{240 \text{ K}}{150 \text{ K}}$$

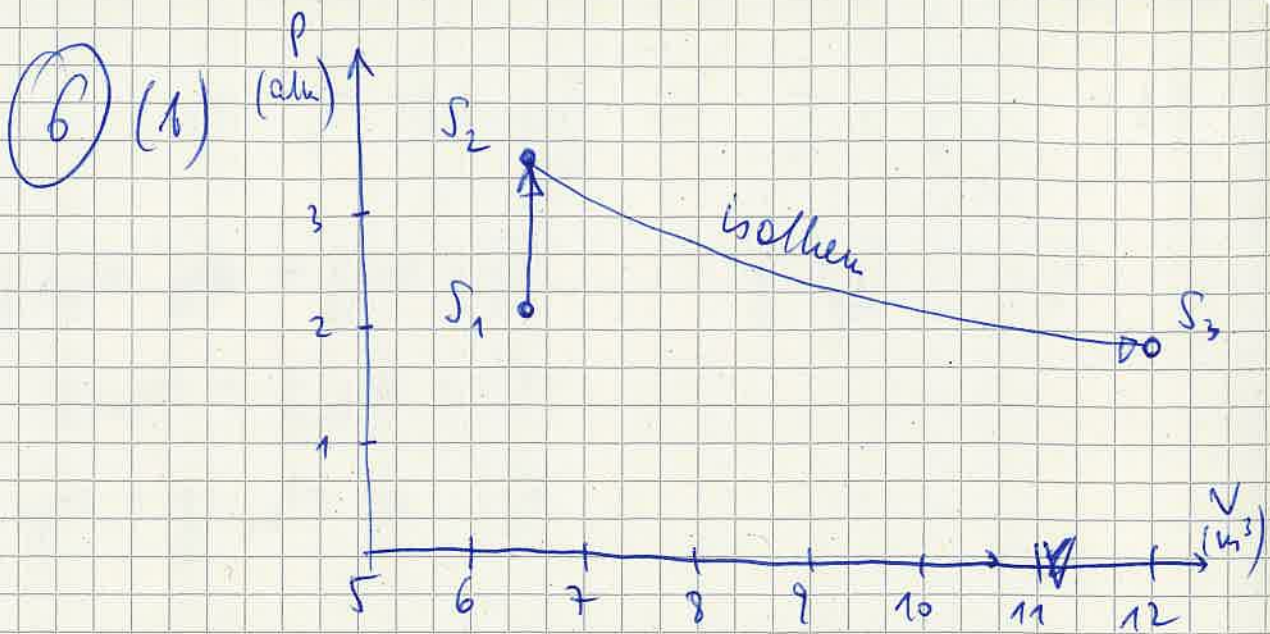
$$p_2 = 3,52 \text{ atm}$$

$S_2 \rightarrow S_3$ (isother): $p \cdot V = \text{const}$

$$p_2 V_2 = p_3 V_3 \rightarrow p_3 = p_2 \cdot \frac{V_2}{V_3} = 3,52 \text{ atm} \cdot \frac{6,5 \text{ m}^3}{12 \text{ m}^3}$$

$$p_3 = 1,906 \text{ atm}$$

$$p_3 = 1,9 \text{ atm}$$



⑦

$$r = 145 \text{ cm} = 1,45 \text{ m} \rightarrow V = \frac{4}{3} r^3 \pi = 12,7700507 \text{ m}^3$$

$$M_{\text{H}_2} = 2,02 \text{ g/mol}$$

(a)

$$T = 25^\circ \text{C} = 298,15 \text{ K}$$

$$p = 2 \text{ bar} = 2 \cdot 10^5 \text{ Pa}$$

$$pV = nRT \rightarrow n = \frac{pV}{RT}$$

$$n = \frac{2 \cdot 10^5 \text{ Pa} \cdot 12,7700507 \text{ m}^3}{8,31446 \frac{\text{J}}{\text{molK}} \cdot 298,15 \text{ K}} = 1030,276397 \text{ mol}$$

$$\underline{n = 1,03 \text{ kmol}}$$

$$(1b) \quad F_A > F_G \rightarrow \text{Grenzfall } F_A = F_G$$

$$F_A = V \cdot \rho_{\text{Luft}} \cdot g$$

$$F_G = m \cdot g + m_{\text{rel}} \cdot g$$

$$F_G = m_{\text{H}_2} \cdot g + m_{\text{rel}} \cdot g$$

$$F_G = n \cdot M_{\text{H}_2} \cdot g + m_{\text{rel}} \cdot g$$

$$\rightarrow n \cdot M_{\text{H}_2} \cdot g + m_{\text{rel}} \cdot g = V \cdot \rho_{\text{Luft}} \cdot g$$

$$m_{\text{rel}} = V \cdot \rho_{\text{Luft}} - n \cdot M_{\text{H}_2}$$

$$= 12,7700105 \text{ m}^3 \cdot 1,22 \frac{\text{kg}}{\text{m}^3} - 1030,276395 \text{ mol} \cdot 2,02 \frac{\text{g}}{\text{mol}}$$

$$= 15,57946 \text{ kg} - 2,081158 \text{ kg} = 13,498304 \text{ kg}$$

$$m_{\text{rel}} = \underline{\underline{13,4 \text{ kg}}}$$

(abunden! damit der
Ballon aufsteigt)

8

$$n = 2,12 \text{ mol}$$

$$T = 25^\circ\text{C} = 298,15 \text{ K}$$

$$S_1: p_1 = 2,71 \text{ bar} = 2,71 \cdot 10^5 \text{ Pa}$$

$T = \text{const}$

$$S_2: p_2 = 7,6 \text{ bar}$$

$$(a) \quad V_1 = \frac{nRT}{p_1} = \frac{2,12 \text{ mol} \cdot 8,31446 \frac{\text{J}}{\text{molK}} \cdot 298,15 \text{ K}}{2,71 \cdot 10^5 \text{ Pa}}$$

$$V_1 = 0,01939257 \text{ m}^3$$

$$pV = \text{const} \rightarrow p_1 V_1 = p_2 V_2$$

$$V_2 = V_1 \frac{p_1}{p_2} = 0,01939257 \text{ m}^3 \cdot \frac{2,71}{7,6}$$

$$V_2 = 0,00691498 \text{ m}^3$$

$$W = - \int_{V_1}^{V_2} p(V) dV = - nRT \cdot \ln \frac{V_2}{V_1}$$

$$0,00691498$$

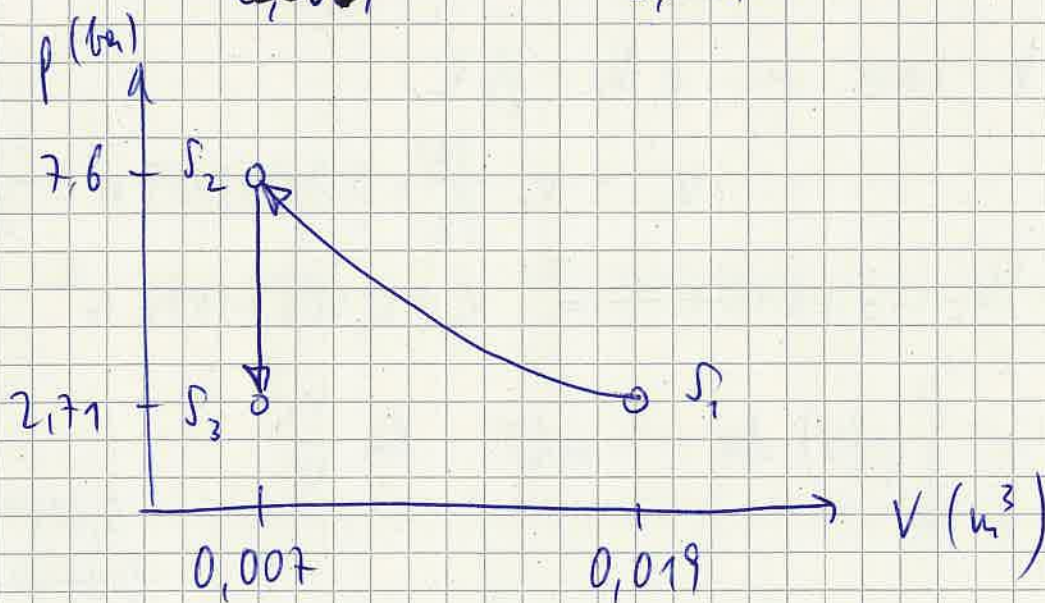
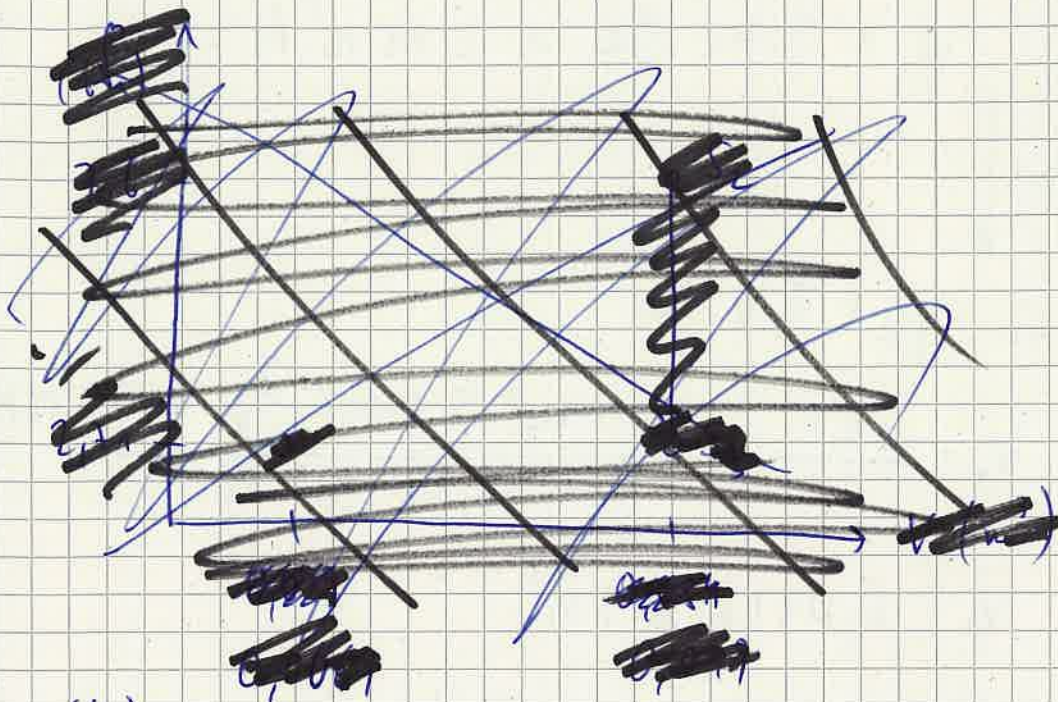
$$= - 2,12 \text{ mol} \cdot 8,31446 \frac{\text{J}}{\text{molK}} \cdot 298,15 \text{ K} \cdot \ln \frac{0,00691498 \text{ m}^3}{0,01939257 \text{ m}^3}$$

$$W = 5419,3533 \text{ J}$$

$$\underline{W = 5,4 \text{ kJ}}$$

$$(b) \quad p_3 = 2,71 \text{ bar}$$

$$V_3 = V_2 = \text{[redacted]} \quad 0,00691498 \text{ m}^3$$



$$(c) \quad pV = nRT$$

$$T_3 = \frac{p_3 V_3}{nR} = \frac{2,71 \cdot 10^5 \text{ Pa} \cdot 0,00691498 \text{ m}^3}{2,12 \text{ mol} \cdot 8,31446 \frac{\text{J}}{\text{mol K}}}$$

$$T_3 = 106,314 \text{ K}$$

$$\underline{T_3 = 1,1 \cdot 10^2 \text{ K}}$$